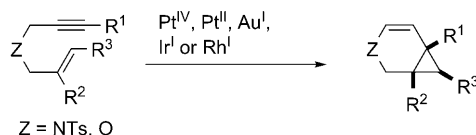


Chiral Tetrafluorobenzobarrelene Ligands for the Rhodium-Catalyzed Asymmetric Cycloisomerization of Oxygen- and Nitrogen-Bridged 1,6-Enynes**

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The transition-metal-catalyzed cycloisomerization reaction of 1,*n*-enynes has emerged as an atom-economical process for the preparation of diverse cyclic compounds.^[1] Nitrogen- and oxygen-bridged enynes are useful starting materials for the construction of heterocyclic building blocks. Since Blum et al. reported the PtCl₄-catalyzed transformation of allyl propargyl ethers into 3-oxabicyclo[4.1.0]heptenes in 1995,^[2] this type of 6-*endo-dig* cycloisomerization has been developed by the use of π -acidic metal^[3] catalysts such as platinum^[4,5] and gold^[6] (Scheme 1). The reaction is thought to be promoted by the

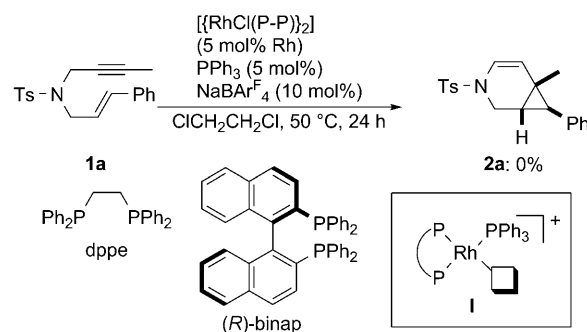


Scheme 1. Cycloisomerization of 1,6-enynes.

coordination of the alkyne moiety on the enyne to the π -acidic metal center. For less π -acidic metals, such as rhodium^[7] and iridium,^[8] their high catalytic activities are induced by the use of CO as a ligand and/or by performing the reaction under a CO atmosphere. In spite of its high synthetic utility for the construction of chiral cyclic skeletons as well as chiral cyclopropanes, there have been only a few reports on the asymmetric variant of the 6-*endo-dig* cycloisomerization reaction of 1,6-enynes to afford bicyclo[4.1.0]heptene derivatives.^[5,8,9] Shibata and co-workers reported the asymmetric cycloisomerization of nitrogen-bridged 1,6-enynes, catalyzed by an iridium-bis(phosphine) complex under CO.^[8] Recently, Marinetti et al. reported high enantioselectivities from the use of a chiral platinum-mono-phosphine complex.^[5c] Herein, we report the asymmetric cycloisomerization of 1,6-enynes to afford 3-aza- and 3-

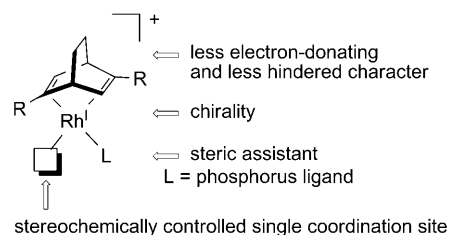
oxabicyclo[4.1.0]heptene derivatives with high enantioselectivity using a rhodium-chiral diene^[10] complex that has an achiral monophosphine as the second ligand.

Based on the reported use of active platinum catalysts containing a single vacant coordination site in the cycloisomerization of dienes^[11] or enynes,^[5] we tested the catalytic activity of a cationic rhodium complex that was coordinated by one bidentate phosphine ligand and one triphenylphosphine ligand for the generation of a cationic rhodium species **I** in the cycloisomerization of a nitrogen-bridged 1,6-enyne **1a** (Scheme 2). However, the use of bis(phosphine) ligands dppe



Scheme 2. Rhodium/phosphine catalysts in the cycloisomerization of 1,6-enyne **1a**. P–P = bis(phosphine) ligand.

and binap did not afford any bicyclo[4.1.0]heptene **2a** product. Next, we focused on the use of chiral diene ligands^[10] for the rhodium-catalyzed asymmetric cycloisomerization of 1,6-enynes, based on our catalyst design that involves three characteristic features (Scheme 3): 1) the coordination of one phosphorus ligand to a cationic rhodium center that is also coordinated by a diene ligand provides a single vacant site at the square-planar rhodium(I) complex; this inhibits the simultaneous coordination of an alkyne and an alkene



Scheme 3. Catalyst design for the rhodium-catalyzed asymmetric cycloisomerization of 1,6-enynes.

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moiety of the 1,6-enyne;^[5,11] 2) the less hindered and less electron-donating character of a diene ligand than a bis(phosphine) ligand is expected to stabilize a rhodium complex that is coordinated by a monophosphine ligand; and 3) the C_2 -symmetric chiral diene ligand offers a good chiral environment.

To estimate the catalytic activity of our rhodium catalyst, the reaction of 1,6-enyne **1a** was carried out in the presence of $[\{\text{RhCl}(\text{diene})\}_2]$ (5 mol % of Rh), PPh_3 (5 mol %), and $\text{NaBAR}^{\text{F}}_4$ (10 mol %; $\text{Ar}^{\text{F}} = 3,5$ -bis(trifluoromethyl)phenyl) in 1,2-dichloroethane at 50 °C for 24 hours (Table 1). The

Table 1: Rhodium-catalyzed cycloisomerization of **1a**.^[a]

Entry	Diene	PR'_3	Yield [%] ^[b]
1	cod	PPh_3	9
2	nbd	PPh_3	7
3	bod	PPh_3	23
4	tfb	PPh_3	58
5	tfb	-	2 ^[c]
6 ^[d]	tfb	-	0
7	(<i>R,R</i>)- L1	PPh_3	48 (50% <i>ee</i>) ^[e]
8	(<i>R,R</i>)- L2	PPh_3	64 (56% <i>ee</i>) ^[e]
9	(<i>R,R</i>)- L3	PPh_3	94 (80% <i>ee</i>) ^[e]
10	(<i>R,R</i>)- L3	$\text{P}(4\text{-MeC}_6\text{H}_4)_3$	94 (80% <i>ee</i>) ^[e]
11	(<i>R,R</i>)- L3	$\text{P}(4\text{-CF}_3\text{C}_6\text{H}_4)_3$	19 (57% <i>ee</i>) ^[e]
12	(<i>R,R</i>)- L3	PCy_3	2
13	(<i>R,R</i>)- L3	$\text{P}(\text{O}-i\text{Pr})_3$	7

[a] Reaction conditions: enyne **1a** (0.10 mmol), $[\{\text{RhCl}(\text{diene})\}_2]$ (2.5 μmol , 5 mol % of Rh), PPh_3 (5.0 μmol), $\text{NaBAR}^{\text{F}}_4$ (0.010 mmol), 1,2-dichloroethane (0.2 mL) at 50 °C for 24 h. [b] Determined by ^1H NMR spectroscopy. [c] 82% of **1a** was recovered. [d] Performed without $\text{NaBAR}^{\text{F}}_4$. [e] The *ee* was determined by HPLC analysis with a chiral stationary phase column (Chiralpak AD-H). cod: 1,5-cyclooctadiene, nbd: norbornadiene, bod: bicyclo[2.2.2]octa-2,5-diene, tfb: tetrafluorobenzobarrelene, Cy: cyclohexyl.

use of rhodium complexes coordinated by cod (1,5-cyclooctadiene), nbd (2,5-norbornadiene), and bod (bicyclo[2.2.2]octa-2,5-diene)^[12] gave 9%, 7%, and 23% yield of cycloisomerization product **2a**, respectively (Table 1, entries 1–3); tfb (tetrafluorobenzobarrelene)^[13] displayed a higher catalytic activity to give **2a** in 58% yield (Table 1, entry 4). The high catalytic activity of the tfb ligand, which has a similar bicyclo[2.2.2]octadiene skeleton to bod, is probably due to its high π -accepting ability.^[14] Reaction in the absence of triphenylphosphine gave a low yield of **2a**, which indicates that a single coordination site at rhodium is essential for the selective formation of the cycloisomerization product (Table 1, entry 5). A neutral complex $[\{\text{RhCl}(\text{tfb})\}_2]$ also had

no catalytic activity, even though it could provide a single coordination site as a monomeric chlororhodium species (Table 1, entry 6). Next, we examined several C_2 -symmetric tfb ligands^[15] in the reaction of 1,6-enyne **1a** to find a successful catalyst for the enantioselective cycloisomerization reactions (Table 1, entries 7–9). Chiral tfb ligand (*R,R*)-**L1**,^[15b] which has methyl group substituents, gave **2a** in 48% yield, and modest enantioselectivity (50% *ee*; Table 1, entry 7). Ligand **L2**, which has longer alkyl chains (pentyl), gave **2a** in 64% yield and 56% *ee* (Table 1, entry 8). (*R,R*)-**L3**,^[15c] which contains $\text{CH}_2\text{OCH}_2\text{OCH}_3$ substituents, displayed a high catalytic activity and enantioselectivity to afford **2a** in 94% yield and 80% *ee* (Table 1, entry 9). The nature of the phosphorus ligands had a significant influence on both the catalytic activity and enantioselectivity (Table 1, entries 10–13). Tri(*para*-tolyl)phosphine displayed the same catalytic activity and enantioselectivity as PPh_3 (Table 1, entry 10). However, the use of electron-deficient tris(4-trifluoromethylphenyl)phosphine afforded **2a** in low yield (19%) and 57% *ee* (Table 1, entry 11). Both tricyclohexylphosphine (Table 1, entry 12) and triisopropyl phosphite (Table 1, entry 13) gave very low yields of **2a**. The relative and absolute configurations of **2a** that was obtained with (*R,R*)-**L3** were determined to be 1*S*,6*R*,7*R* by X-ray crystallographic analysis (Figure 1).^[16]

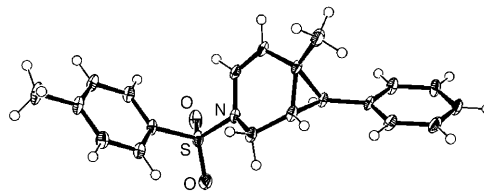


Figure 1. ORTEP illustration of **2a**; thermal ellipsoids are drawn at the 50% probability level.

Table 2 summarizes the results obtained for the cycloisomerization reactions of nitrogen-bridged 1,6-enynes **1**. The reaction was carried out using $[\text{RhCl}(\text{PPh}_3)(\text{R,R})\text{-L3}]$ (5 mol % of Rh) as a precursor to the active catalytic species that possesses one vacant coordination site, which was prepared by the reaction of $[\{\text{RhCl}(\text{R,R})\text{-L3}\}_2]$ with PPh_3 in dichloromethane. The asymmetric cycloisomerization of 1,6-enynes **1a–1e**, which have aryl groups on the alkene moiety, proceeded to give the corresponding bicyclic compounds **2a–2e** in good yields,^[17] and enantioselectivities in the range 76–95% *ee* (Table 2, entries 1–5). Enynes containing *ortho* substituents on the benzene ring, 2-MeO (**1c**) and 2,6-Me₂ (**1d**), increased the enantioselectivity (90 and 95% *ee*, respectively; Table 2, entries 3 and 4). 1,6-Enyne **1f**, which contains a trifluoromethanesulfonyl substituent on the nitrogen atom, also gave cycloisomerization product **2f** in 73% yield and 73% *ee* (Table 2, entry 6). The reaction of 1,6-enyne **1g**, which has a methallyl group, proceeded smoothly under mild conditions, although the enantioselectivity of the product **2g** was moderate (68% *ee*; Table 2, entry 7).

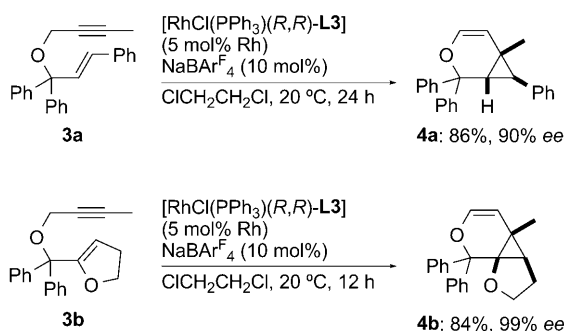
The same type of cycloisomerization was observed for oxygen-bridged 1,6-enynes [Eqs. (1) and (2)]. Thus, the

Table 2: Rhodium-catalyzed cycloisomerization of 1,6-enyne **1**.^[a]

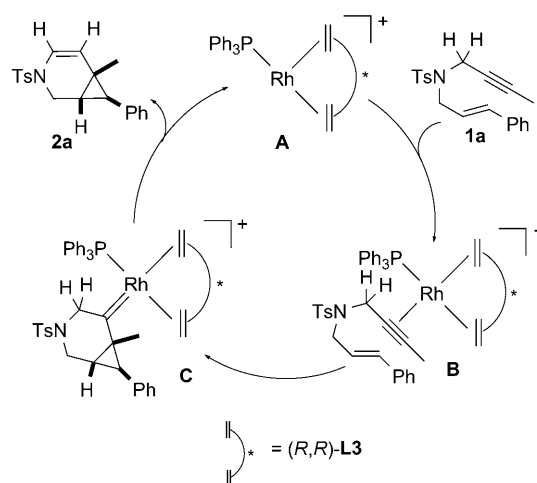
$\text{X-N} \begin{array}{c} \text{---} \text{C} \equiv \text{C} \text{---} \\ \\ \text{R}^1 \end{array} \text{---} \text{CH} \text{---} \text{CH}_2 \text{---} \text{R}^2 \quad \xrightarrow[\text{ClCH}_2\text{CH}_2\text{Cl}]{\begin{array}{c} [\text{RhCl}(\text{PPh}_3)(\text{R},\text{R})\text{-L3}] \\ (5 \text{ mol\% Rh}) \\ \text{NaBARF}_4 (10 \text{ mol\%}) \end{array}} \quad \text{X-N} \begin{array}{c} \text{---} \text{C} \text{---} \text{C} \text{---} \\ \quad \\ \text{R}^1 \quad \text{R}^2 \end{array}$					
1a: X = Ts, R ¹ = H, R ² = Ph	1e: X = Ts, R ¹ = H, R ² = 4-ClC ₆ H ₄				
1b: X = Ts, R ¹ = H, R ² = 4-MeOC ₆ H ₄	1f: X = Tf, R ¹ = H, R ² = Ph				
1c: X = Ts, R ¹ = H, R ² = 2-MeOC ₆ H ₄	1g: X = Ts, R ¹ = Me, R ² = H				
1d: X = Ts, R ¹ = H, R ² = 2,6-Me ₂ C ₆ H ₃					
[Ts = SO ₂ C ₆ H ₄ -4-Me, Tf = SO ₂ CF ₃]					
Entry	Enyne	T [°C]	t [h]	Yield [%] ^[b]	ee [%] ^[c]
1	1a	50	24	94 (2a)	80
2	1b	40	24	91 (2b)	81
3	1c	40	24	87 (2c)	90
4	1d	60	48	71 (2d)	95
5	1e	60	24	92 (2e)	76
6	1f	60	48	73 (2f)	73
7	1g	40	24	81 (2g)	68

[a] Reaction conditions: enyne **1** (0.20 mmol), [RhCl(PPh₃)(R,R)-L3] (0.010 mmol, 5 mol% of Rh), NaBARF₄ (0.020 mmol, 10 mol%), 1,2-dichloroethane (0.4 mL). [b] Yield of isolated product. [c] Determined by HPLC analysis with a chiral stationary phase column (Chiralpak AD-H).

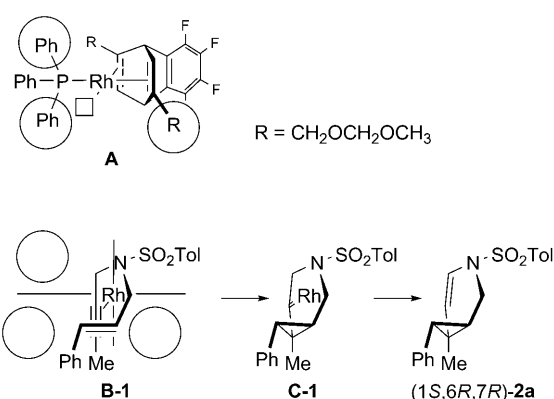
reaction of **3a** gave 3-oxabicyclo[4.1.0]heptene **4a** in 86% yield and 90% ee [Eq. (1)].^[18] Very high enantioselectivity (99% ee) was observed for the reaction of 1,6-enyne **3b**, which afforded tricyclic compound **4b** [Eq. (2)].



It has been proposed that the mechanism for the Pt^{II}-catalyzed cycloisomerization reaction involves a platinum carbenoid intermediate.^[1,4,19] Provided that the mechanism for the π-acidic Rh^I-catalyzed formation of bicyclo[4.1.0]heptenes is similar to that for the Pt^{II}-catalyzed reaction, we suggest that the catalytic cycle of the cycloisomerization proceeds as shown in Scheme 4. The coordination of the alkyne moiety of 1,6-enyne **1a** to coordinatively unsaturated cationic rhodium(I) species **A**, which was generated from [RhCl(PPh₃)(R,R)-L3] and NaBARF₄, affords η²-alkyne–rhodium species **B**. Species **B** undergoes 6-endo-dig cyclization with the formation of two carbon–carbon bonds to form rhodium carbenoid **C**. β-Hydrogen shift affords the product (**2a**) whilst also regenerating the cationic rhodium species (**A**). It is most likely that the configuration of the product (**2a**) is determined during the carbon–carbon bond-formation step on η²-alkyne–rhodium species **B**. The enantioselectivity achieved in this catalytic system can be rationalized using a quadrant diagram (Scheme 5). The ligands on cationic



Scheme 4. Plausible catalytic cycle.



Scheme 5. Proposed stereochemical pathway.

rhodium species **A** effectively shield three quadrants, and the substituents on the tfb ligand shape the less shielded site at the fourth quadrant. The alkene part of the 1,6-enyne approaches the alkyne moiety that has been activated by the π-acidic rhodium center. The 6-endo-dig cyclization step is directed by the steric bulk of the tosyl group, the two proximal phenyl groups of the triphenylphosphine, and the CH₂OCH₂OCH₃ substituent on the (R,R)-L3. Therefore, cyclization from **B-1** (Scheme 5) to give rhodium–carbenoid **C-1** is the most favorable, which leads to (1S,6R,7R)-**2a**.

In summary, we have developed a rhodium-catalyzed asymmetric cycloisomerization reaction of nitrogen- and oxygen-bridged 1,6-enynes that affords 3-aza- and 3-oxabicyclo[4.1.0]heptenes in high yields and high enantioselectivities. The reaction was accomplished using a cationic rhodium complex that was coordinated by PPh₃ and a chiral diene ligand.

Experimental Section

[RhCl(PPh₃)(R,R)-L3] (7.7 mg, 0.010 mmol) and NaBARF₄ (18.4 mg calculated as dihydrate, 0.020 mmol) were added to a solution of 1,6-enyne **1a** (67.9 mg, 0.20 mmol) in 1,2-dichloroethane (0.4 mL), and the mixture was stirred at 50°C for 24 hours. The mixture was passed

through a short column of silica gel with hexane/ethyl acetate/Et₃N (20:5:1) as eluent. After evaporation of the solvent, the residue was subjected to flash column chromatography on silica gel (hexane/ethyl acetate/CHCl₃/Et₃N=20:1:1:1) to give **2a** (63.6 mg, 0.19 mmol, 94 %).

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- [16] CCDC 753308 (**2a**) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [17] The reaction of a terminal alkyne analogous to **1a** gave only 4 % yield of cycloisomerization product **2**.
- [18] The use of an oxygen-bridged 1,6-enyne that has no phenyl groups in the allylic position gave a low yield of the cycloisomerization product (18 % yield for 2-butynyl (*E*)-3-phenyl-2-propenyl ether). The reaction of a 1,6-enyne, bridged by carbon atoms, did not give the corresponding bicyclo[4.1.0]heptene derivative.
- [19] For examples of theoretical studies of Pt^{II}-catalyzed cycloisomerization, see: a) C. Nevado, D. J. Cárdenas, A. M. Echavarren, *Chem. Eur. J.* **2003**, *9*, 2627; b) E. Soriano, P. Ballesteros, J. Marco-Contelles, *J. Org. Chem.* **2004**, *69*, 8018; c) E. Soriano, P. Ballesteros, J. Marco-Contelles, *Organometallics* **2005**, *24*, 3172; d) E. Soriano, P. Ballesteros, J. Marco-Contelles, *Organometallics* **2005**, *24*, 3182; e) E. Soriano, J. Marco-Contelles, *Acc. Chem. Res.* **2009**, *42*, 1026.